Stratospheric and Mesospheric HO_x : Results from Aura MLS and FIRS-2

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- 1 Observations of OH and HO₂ from Aura MLS for four seasons and diur-
- 2 nal profiles from the FIRS-2 balloon instrument for Fall 2004 are compared
- 3 with photochemical model simulations testing three sets of kinetics param-
- 4 eters. MLS and FIRS-2 OH profiles, between 25-60 km, are lower than model
- 5 results using standard kinetics. Use of a faster, previously published rate con-
- 6 stant for O+OH leads to better agreement with MLS and FIRS-2 profiles
- 7 of OH. A 20% increase in the rate of HO_2+OH and the faster rate for O+OH
- 8 results in improved overall agreement with observations of OH, HO_2 , HO_x ,
- 9 and HO_2/OH . Since the MLS and FIRS observations of HO_x are reasonably
- well described by these models, they are therefore not consistent with the
- 11 previously reported HO_x dilemma. However, all models considered here re-
- 12 sult in calculated odd oxygen loss exceeding production, consistent with the
- 13 long standing ozone deficit problem.

1. Introduction

Simultaneous observations of OH and HO₂ from the Microwave Limb Sounder (MLS) instrument on board the Aura satellite, launched July 15, 2004, provide a unique opportunity to test our understanding of stratospheric HO_x (OH+HO₂). We present an analysis of daytime zonally averaged profiles of OH and HO₂ for four seasons using a photochemical model constrained by MLS observations of HO_x precursors. Measurements of OH and HO₂ acquired by the Far-Infrared Spectrometer (FIRS-2) instrument during an Aura validation balloon campaign in September 2004 are also examined.

In the stratosphere, OH is primarily produced through the reaction of water with metastable oxygen, O(1 D), and by water photolysis above 60 km. HO_x is lost primarily through the reaction

$$HO_2 + OH \rightarrow H_2O + O_2. \tag{1}$$

Previous observations of either OH or HO_2 alone have shown poor agreement with model simulations. As a result, two studies have suggested modifications to the rate constants of HO_x partitioning reactions to reach better agreement between measurements and model calculations (see auxiliary material for further discussion¹). Ground based microwave observations of HO_2 were used to suggest a 60-80% decrease in the rate of

$$HO_2 + O \rightarrow OH + O_2$$
 (2)

[Clancy et al., 1994]. Mesospheric OH observations by the Middle Atmospheric High Resolution Spectrograph Investigation (MAHRSI) instrument in November 1994 led to the suggestion of either a 50% reduction in the rate of reaction (2) or both a 20% reduction of rate (2) and a 30% increase in rate (1) [Summers et al., 1997]. However, without simul-

- 25 taneous observations of OH and HO₂, it is difficult to attribute the above discrepancies
- 26 to HO_x loss, production, or partitioning.
- 27 These suggested changes were tested against an OH profile measured by MAHRSI in
- 28 August 1997 [Conway et al., 2000]. Their results showed that the kinetic changes needed to
- 29 match the MAHRSI OH profile above 50 km led to poorer agreement between modeled and
- 30 measured OH from 35-45 km. No particular kinetics change allows models to reproduce
- 31 MAHRSI OH in both the mesosphere and the upper stratosphere. This is known as the
- 32 " HO_x dilemma" [Conway et al., 2000].
- Loss of odd-oxygen $(O_x=O_3+O)$ is dominated by HO_x catalytic processes above 45 km.
- 34 Ozone, the main component of O_x at these altitudes, should be in photochemical steady
- state. However, calculated loss of O_x generally exceeds production by $\sim 35\%$ [e.g. Jucks et
- 36 al., 1996, Osterman et al., 1997. This leads to an underprediction of upper stratospheric
- 37 O_3 , commonly known as the "ozone deficit problem".
- 38 The kinetics changes suggested by Clancy et al. [1994] and Summers et al. [1997] lead
- 39 to good agreement with measured mesospheric HO₂ and OH, respectively, and also largely
- 40 resolve the ozone deficit problem. These results are driven by a reduction in the rate of
- 41 (2), resulting in more HO_2 , less OH, and slower O_x removal compared to a standard
- 42 model. In contrast, Jucks et al. [1998] suggested the rates of reactions (1) and (2) must
- 43 both be reduced by 25% to best explain FIRS-2 observations of OH and HO_2 . The Jucks
- 44 et al. [1998] kinetics change has a negligible effect on the ozone deficit problem. Below,
- 45 we investigate the implications of recent Aura MLS and FIRS-2 observations for the HO_x
- 46 dilemma and ozone deficit problem.

2. Measurements and Model

- The Aura MLS instrument measures OH at 2.5 THz and HO₂ at 643 GHz [Pickett,
- 48 2006]. Validation of MLS OH and HO₂ by comparison with balloon-borne remote sensing
- 49 measurements of these species is described by *Pickett et al.* [2006].
- The MLS profiles presented here are 15 day zonal averages, spanning 34±10°N, centered
- on September 23 (fall) and December 23, 2004 (winter) as well as March 15 (spring) and
- 52 June 15, 2005 (summer), based on version 1.51 of the retrieval software. The local solar
- 53 time (LST) of the observations is ~ 13.30 hr. MLS observations of OH, HO₂, HO_x, and
- 54 HO₂/OH are shown in Figure 1. Here, we only consider data below 60 km, because above
- 55 60 km only observations of OH are available [Pickett et al., 2006]. Precision in the 15
- 56 day averages for OH and HO₂ is good with negligible uncertainty. The error bars in
- 57 Figures 1a and 1b are equal to 10%, which represents our estimate of the uncertainty
- 58 in instrument calibration (i.e., measurement accuracy) [Pickett et al., 2006]. Raw MLS
- 59 HO₂ profiles (not shown) exhibit oscillatory behavior that is likely a retrieval
- 60 artifact [Pickett et al., 2006]. This behavior will result in non-zero reduced chi
- square (χ^2_r) values even for a model that simulates quite well the overall shape
- and magnitude of HO_x species. To avoid this situation, we have smoothed the
- 63 raw MLS profiles of HO₂ using a boxcar average (see auxiliary material ¹) to
- arrive at the HO_2 profiles used throughout.
- 65 Observations from FIRS-2 were taken by a thermal emission far-infrared Fourier trans-
- 66 form spectrometer [Jucks et al., 1998] on board a balloon gondola launched from Ft.
- 67 Sumner, NM $(34.5^{\circ}\text{N}, 104^{\circ}\text{W})$ on September 23, 2004. These profiles are from ~ 1 hour

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limb scans. Seven profiles, taken over the course of the day (7:30-17:00 LST), are used 68 for statistical comparison. We consider all data below the balloon float altitude (38 km) 69 and three points above (42, 44 and 48 km) to account for the poorer vertical resolution 70 above float. The OH and HO₂ error bars are the root sum squared (RSS) combination 71 of 1σ estimates of accuracy and precision. Error bars for both HO_x and HO_2/OH shown **72** in Figures 1c and 1d are the RSS propagation of the errors in OH and HO₂ from the **73** respective instruments. 74 The photochemical model is constrained by MLS measurements of H₂O, O₃, N₂O, CO, 75 and temperature for each season. The model assumes a balance of production and loss **76** for each species integrated over 24 hrs and has been used in previous studies to analyze 77 observations from balloon, satellite, and aircraft platforms [e.g., Pickett et al., 2006, Jucks **78** et al., 1998]. Profiles of Cl_y , NO_y , and CH_4 are specified using well established tracer-**79** tracer relations [Jucks et al., 1998]. The model includes mesospheric chemistry and solar 80 81 cycle effects, a new feature described in the auxiliary material of *Pickett et al.* [2006]. We show model results for several sets of kinetic parameters: a) JPL02 kinetics [Sander 82 et al., 2003] (hereafter Mdl_{JPL02}); b) same as JPL02 except the Smith and Stewart [1994] 83 (hereafter SmSt94) rate constant for O+OH (Mdl_{SmSt}) c) same as Mdl_{SmSt}, except a 20% 84 increase in the rate of HO_2+OH (Mdl_C). The SmSt94 rate constant for O+OH is $\sim 20\%$ 85 faster than the JPL02 rate and is within the JPL02 uncertainty (Figure 2). The suggested 86 increase in the rate of O+OH is consistent with Jucks et al. [1998], who suggested a 87

reduction in $k(O+HO_2)/k(O+OH)$. Two recent laboratory studies of the O+OH rate

constant that report contrasting results, published as our work was being completed, are
discussed in the auxiliary material¹.

3. Results and Discussion

The MLS OH profiles (Figure 1a) all peak near 45 km. Differences in peak values are 91 due to seasonal changes in solar declination. The MLS OH observations and Mdl_{JPL02} 92calculations result in χ^2_r =12.3, between 25-60 km and considering all seasons (see auxil-93 iary material¹ for description of χ^2_r ; a value of 1 indicates that model profiles generally 94 lie within measurement uncertainty). Mdl_{JPL02} overestimates observed OH between 40-60 95 km, often outside of the measurement uncertainty. Better agreement between modeled 96 and measured OH is achieved for Mdl_{SmSt} . This comparison results in a $\chi^2_r=3.1$ for OH. 97 Results for Mdl_C , described below, lead to a χ^2_r =1.6 (Figure 3). The good agreement 98 between measured OH profiles and the Mdl_C simulation at all altitudes and seasons indi-99 cates that MLS observations do not exhibit a " HO_x dilemma" as reported by Conway et 100 al. [2000] for MAHRSI observations of OH. 101 The closest FIRS-2 observations in time (LST=13.6 hr) to the MLS overpass are shown 102 in Figure 1a. The $\chi^2_{\rm r}$ between FIRS-2 observations of OH and the three model cases are larger than for the MLS comparison. The χ^2_r values for FIRS-2 are 16.7 for Mdl_{JPL02}, 104 11.8 for Mdl_{SmSt}, and 10.7 for Mdl_C (Figure 4; profiles at seven times have been used to 105 calculate $\chi^2_{\rm r}$, as described above). These larger values are due to the influence of the 106 higher altitude measurements of OH, which are much smaller than model values. The 107 sense of the discrepancies between FIRS-2 OH and the Mdl_C calculation for September 108 2004 at various altitudes is the same as noted by Conway et al. [2000]. However, the 109

- 110 FIRS-2 discrepancies are smaller, particularly near 40 km. The Mdl_C simulation provides
- a reasonably good description of the shape and abundance of the FIRS-2 OH profile.
- Hence, the FIRS-2 observations are also not consistent with a HO_x dilemma.
- Figure 1b shows comparisons of measured and modeled HO₂. Mdl_{JPL02} overestimates
- 114 MLS HO_2 mainly below 40 km, resulting in a χ^2_r =2.9. In contrast to the compari-
- son for OH, Mdl_{SmSt} results in a slightly higher value of χ^2_r (4.1) than Mdl_{JPL02} . Best
- agreement with MLS HO₂ is achieved by Mdl_C, with $\chi^2_r=1.7$. The HO₂ profile measured
- by FIRS-2 is generally higher than observed by MLS (Figure 1b). All three models give
- 118 excellent agreement with FIRS-2 HO₂ (Figure 4).
- We have determined, through a series of sensitivity studies, that a reasonably good
- overall description of measured OH, HO_2 , HO_x , and HO_2/OH is achieved using Mdl_C ,
- which includes a 20% increase in (1) and the SmSt94 rate for O+OH. Profiles of total
- 122 HO_x and HO_x partitioning from MLS and FIRS-2 are shown in Figures 1c and 1d. The
- 123 MLS profiles are affected by the oscillations in HO₂. Best agreement between measured
- and modeled MLS HO_x is found for Mdl_C, with a χ^2_r =3.0 (Figure 3). For HO_x partition-
- ing (e.g., HO₂/OH) measured by MLS and FIRS-2, the two simulations using the SmSt94
- 126 rate for O+OH result in slightly better agreement compared to Mdl_{JPL02} (Figures 3 and
- 127 4). Considering the suite of model and measured OH, HO_2 , HO_x , and HO_x partitioning,
- 128 represented by "Total" in Figures 3 and 4, Mdl_C kinetics leads to the best overall simula-
- 129 tions for both the MLS (χ^2_r =2.3) and FIRS-2 (χ^2_r =7.5) data sets, for the suite of model
- 130 simulations considered here.

If we use the JPL02 rate for O+OH, no simple change to the rate of HO_2+OH improves the simulation of both MLS OH and HO_x in a manner comparable to Mdl_C kinetics. Likewise, it is difficult to reconcile the observations and model results considering only uncertainty in the O+OH rate. The range of model calculations based on the JPL02 uncertainty in O+OH is given in the auxiliary material¹.

4. Ozone Deficit Problem

Calculated (O_x) production (P) and loss (L) rates during Fall 2004, for the three model 136 runs, are shown in Figure 5. Production is the same for all cases because model O₃ has 137 been constrained to the daytime MLS profile. Calculated L exceeds P throughout the 138 upper stratosphere and lower mesosphere, consistent with an ozone deficit problem. 139 Introduction of the SmSt94 rate for O+OH leads to an increase in calculated HO₂ 140 compared to Mdl_{JPL02}. This increased HO₂ results in larger L-P compared to Mdl_{JPL02} 141 because $O+HO_2$ is a rate determining step of O_x loss. The Mdl_C simulation results in 142 a value of L-P that is intermediate between the other two simulations above 50 km: the 143 increase in HO_2+OH results in lower HO_x and hence slower O_x loss by all HO_x cycles 144 compared to the Mdl_{SmSt} simulation. A 50% reduction in the rate of O+HO₂ results in balance of P and L near 40 km, as suggested by Summers et al. [1997], but leads to poorer 146 agreement with MLS and FIRS-2 HO_x profiles compared to the other simulations shown above (see auxiliary material¹). 148 There have been many suggestions in the literature regarding possible resolutions to the 149 HO_x dilemma and the O_3 deficit problem. It has been suggested that reactions involving vibrationally excited $O_2(\nu \geq 26) + O_2$ could solve the ozone deficit problem by providing an autocatalytic source of O_x [Miller et al., 1994]. However, Slanger and Copeland [2003] question the existence of this reactive pathway. Varandas [2004] suggested reactions involving vibrationally excited O_2 and OH could be important for both the HO_x dilemma and the O_3 deficit problem. However, Smith and Copeland [2005] have raised doubts regarding the suggestion of Varandas [2004]. Our observations and simulations, taken at face value, suggest a continued need to resolve the ozone deficit problem without recourse to major perturbations in the kinetic parameters that regulate HO_x .

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¹Auxiliary material is available via Web browser or via Anonymous FTP from ftp://ftp.agu.org/apend/

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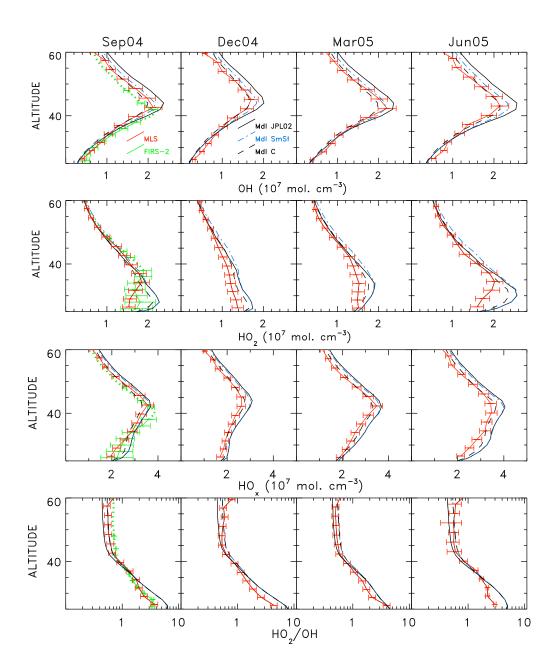


Figure 1. 1st Row: MLS OH profiles (red curve) for four seasons and model results for a) JPL02 kinetics, "Mdl_{JPL02}" (black solid), b) Smith and Stewart [1994] rate for O+OH, "Mdl_{SmSt}" (blue dashed dot), c) Smith and Stewart [1994] rate for O+OH and a 20% increase to OH+HO₂, "Mdl_C" (see text) (black dashed), FIRS-2 observations from Sept 23, 2004 (green curve, data fit to assumed profile shape above float altitude indicated by green dotted curve) are also shown. 2nd Row: same as top row except for HO₂; 3rd Row: same as top row except for HO₂, 4th Row: same as top row except for HO₂/OH.

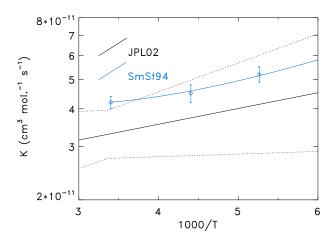


Figure 2. O+OH reaction rate from JPL02 (black) and from *Smith and Stewart* [1994] (blue). Black dotted curves denote uncertainties from JPL02.

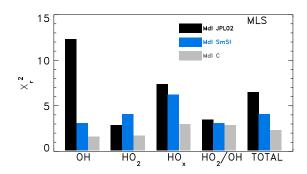


Figure 3. $\chi^2_{\rm r}$ between MLS measurements and: "Mdl_{JPL02}" (black), "Mdl_{SmSt}" model (blue solid), "Mdl_C" (see text) (gray), for OH, HO₂, HO_x, and HO₂/OH. Total represents average of $\chi^2_{\rm r}$ for other 4 parameters.

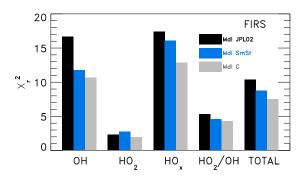


Figure 4. Same as Figure 3, except for comparison of models with FIRS-2 observations

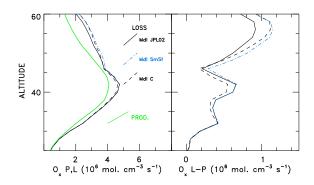


Figure 5. Left Panel: Fall 2004 production (green curve) and loss of odd oxygen from Mdl_{JPL02} (black curve), Mdl_{SmSt} (blue dashed dot), and Mdl_{C} (black dashed), Right panel: Loss-Production of O_x for the 3 scenarios shown in the left panel.

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O+HO₂ Reaction Kinetics

Early studies suggested a large decrease in the rate of the O+HO₂ reaction was needed to achieve good agreement between either HO₂ or OH and model calculations [e.g., Clancy et al., 1994, Summers et al., 1997]. The effects of a 50% reduction to the O+HO₂ reaction rate, as suggested by Summers et al. [1997] are shown in Figure 6 (Mdl_D). Above 40 km, Mdl_D results in lower OH and higher HO₂ than MLS observations. Below 45 km, Mdl_D OH is lower than than observations from FIRS-2. HO₂/OH calculated from Mdl_D is generally higher than all observations (including errors). A comparison of χ^2_r for Mdl_D (versus MLS data) to values found using MDL_{JPLO2} and MDL_{SmSt} is shown in Figure 7. Clearly, MDL_D is less consistent with the MLS observations of HO_x than the other simulations.

As stated in the manuscript, a 50% decrease of the $O+HO_2$ reaction rate does lead to a balance of production and loss near 40 km even though though the calculated profiles of HO_x are not in good agreement with observations (Figure 8).

Box Car Averaging of HO₂ Profiles

As stated in the manuscript, the MLS profiles of HO_2 are smoothed to avoid the oscillatory behavior that is apparent in the HO_2 retrievals shown by Pickett et al. [2006]. Smoothed profiles of HO_2 allow for a more straightforward interpretation of the reduced chi squared analysis (see text). We have applied the following boxcar averaging scheme to remove these oscillations.

$$\sum_{i=1}^{n} HO_2(i) = 0.25 \times HO_2(i-1) + 0.5 \times HO_2(i) + 0.25 \times HO_2(i+1),$$
(1)

where i indicates an altitude index. For the HO₂ profile in winter, this box car averaging procedure was applied 3 times to remove an oscillation that was apparent after the previous smoothings. We calculated the endpoints by taking the averaging the endpoint and the next closest point.

O+OH Reaction Kinetics

The range of model calculations based on the recommended JPL02 uncertainties (Sander et al., 2003) from the O+OH reaction is shown in Figure 9 (gray shaded regions). MLS observations of OH above 40 km are often lower than the modeled region though the errors bars overlap. Even with the consideration of the O+OH uncertainties, it is difficult to reconcile the observations with the model results. Calculated $\rm HO_2$ spans the range of $\rm HO_2$ observations from the MLS instrument. We again note that the MLS $\rm HO_2$ profiles exhibit oscillations that may be a retrieval artifact. This behavior is not seen in the FIRS-2 observations. Calculated $\rm HO_x$ (OH+HO₂) and partitioning ratios are affected by the high calculated OH. Modeled $\rm HO_x$ is often higher than observations above 40 km. The modeled partitioning ratio is outside of the FIRS-2 error bars through most of the stratosphere.

Two new laboratory studies of the O+OH rate constant, published as our study was being completed, provide contrasting results. The cold temperature study by Carty et al. [2005] suggests a temperature independent rate for this reaction that is 30-50% lower than the JPL02 recommendation, but still within the JPL02 uncertainty (Figure 10). The rate constant measured by Robertson and G.P. Smith [2006] is faster than the JPL02 rate, with a steeper temperature dependence than was seen in any other study (Figure 10). The Robertson and G.P. Smith [2006] rate is similar to the I.W.M. Smith and Stewart [1994] rate for temperatures between 180 and 230 K. At a temperature of 295 K, the Robertson and G.P. Smith [2006] rate is similar to the JPL02 rate. Use of the Robertson and G.P. Smith [2006] rate constant for O+OH results in values of χ^2_r that lie between values of Mdl_{JPL02} and Mdl_{SmSt}, as would be expected by the behavior of the respective rate constants shown in Figure 10.

The laboratory studies of I.W.M. Smith and Stewart [1994] and Robertson and G.P. Smith measured the decay of OH in excess O, monitored by laser induced fluorescence, using experimental setups that were similar. Carty et al. [2005] (I.W.M. Smith is a co-author of this study) also observed the decay of OH by laser induced fluorescence.

However, Carty et al. [2005] used an experimental setup quite different than the other two studies, involving Laval nozzles to achieve extremely low temperature in supersonic flow. Further discussion of the different experimental methods and results is beyond the scope of this paper.

Reduced X_r^2

For the statistical results presented in this study, we use a reduced X_r^2 analysis, where

$${X_r}^2 = \frac{1}{N} \sum_{i=1}^{N} \left(\frac{Observation(i) - Model(i)}{Overall \ observational \ uncertainty(i)} \right)^2 \tag{2}$$

A value of unity indicates the model profile is generally within the uncertainty of the observations.

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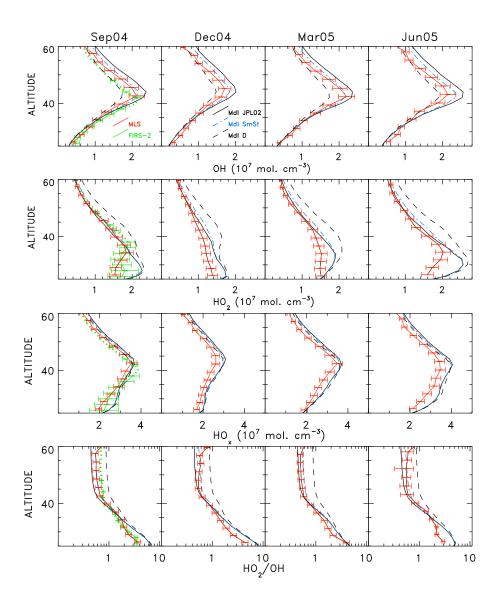


Figure 6. 1st Row: MLS OH profiles (red curve) for four seasons and model results for a) JPL02 kinetics "Mdl $_{\rm JPL02}$ " (black solid), b) reaction rate of O+OH from Smith and Stewart [1994], "Mdl $_{\rm SmSt}$ " (blue dashed dot), c) a 50% reduction in the reaction rate of O+HO $_2$ "Mdl $_{\rm D}$ " (black dashed), FIRS-2 observations from Sept 23, 2004 (green curve, data fit to assumed profile shape above float altitude indicated by green dotted curve) are also shown. 2nd Row: same as top row except for HO $_2$; 3rd Row: same as top row except for HO $_2$, 4th Row: same as top row except for HO $_2$ /OH.

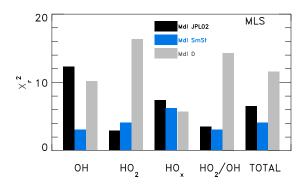


Figure 7. $\chi^2_{\rm r}$ between MLS measurements and: "Mdl_{JPL02}" (black), "Mdl_{SmSt}" model (blue solid), "Mdl_D" (gray), for OH, HO₂, HO_x, and HO₂/OH. Total represents average of $\chi^2_{\rm r}$ for other 4 parameters.

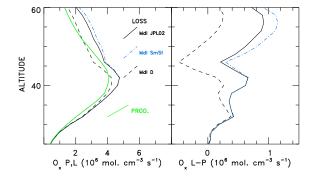


Figure 8. Left Panel: Fall 2004 production (green curve) and loss of odd oxygen from JPL02 kinetics "Mdl_{JPL02}" (black curve), reaction rate of O+OH from Smith and Stewart [1994], "Mdl_{SmSt}" (blue dashed dot), a 50% decrease in O+HO₂, "Mdl_D" (black dashed), Right panel: Loss-Production of O_x for the 3 kinetic scenarios shown in the left panel.

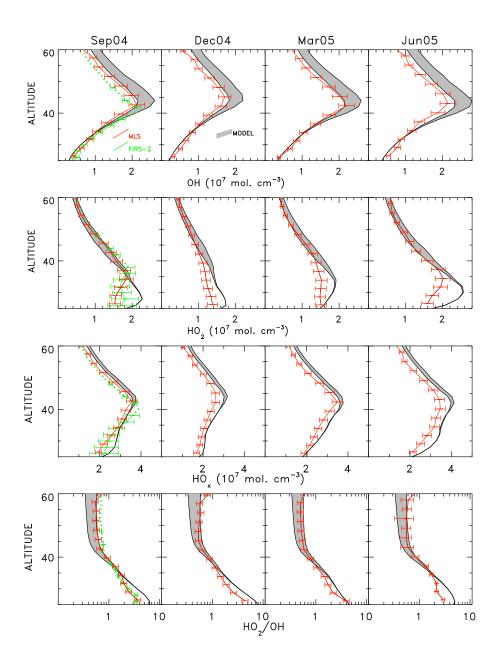


Figure 9. 1st Row: MLS OH profiles (red curve) for four seasons and model results for the recommended JPL02 uncertainty in the O+OH reaction (gray shaded region), FIRS-2 observations from Sept 23, 2004 (green curve) are also shown; 2nd Row: same as top row except for HO₂; 3rd Row: same as top row except for HO_x, 4th Row: same as top row except for HO_x/OH

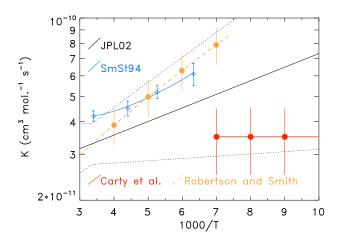


Figure 10. Same as Figure 2 in the main body of the paper, with the addition of rate constants for O+OH measured by Carty et al. [2005] (red) and Robertson and Smith [2006] (orange dashed). Uncertainties for the Robertson and Smith [2006] rate constant represent are 16%, representing the 1σ overall uncertainty stated in their Conclusions section. The uncertainty for Carty et al. [2005] is the value given in their abstract, which applies to all temperatures of the experiment (39-142 K), and represents a combination of the statistical differences of various runs combined with other sources of systematic error.